Nonlinear Attosecond Spectroscopy

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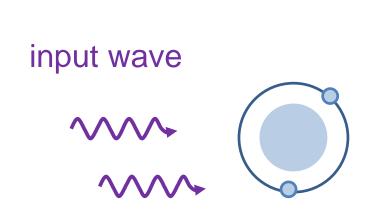
Abstract

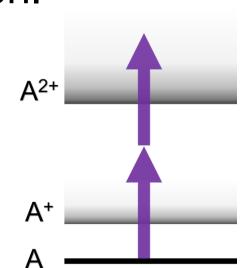
In the project NONLINEAR-ATTO we are developing a system for the investigation of correlated electronic dynamics by nonlinear attosecond extreme ultraviolet (XUV) spectroscopy.

Simple quantum systems will be investigated by two-photon double ionization giving insight to electron-electron correlation and by performing XUV-XUV pump-probe experiments with attosecond time resolution.



When we irradiate an atomic/molecular system with intense light pulses the medium's response becomes nonlinear. Thus, it may absorb two photons and emit a new photon with double frequency. When this absorption leads to ionization it is referred to as two-photon double ionization.







Multiphoton ionization was first proposed by Maria Göppert Mayer, Nobel laureate in 1963, with her dissertation in the University of Göttingen in 1931, focused on 'two quanta jumps'.



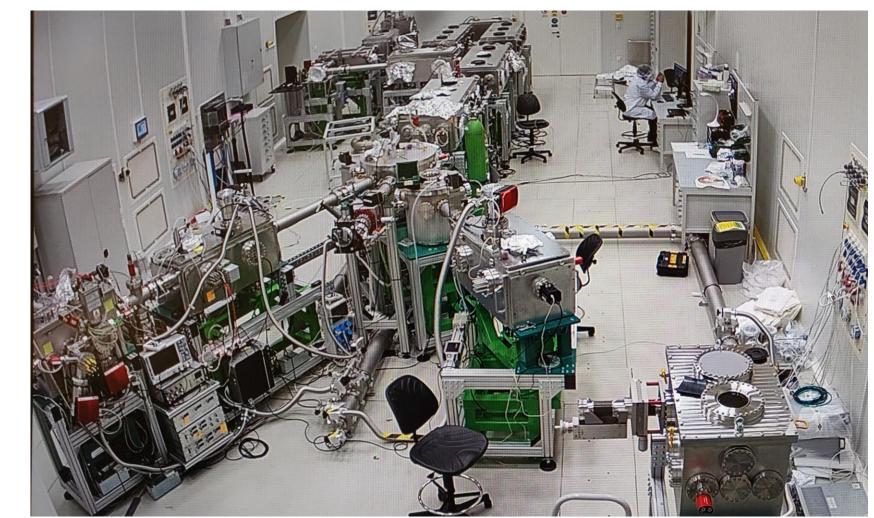
Ann. Phys., Lpz. B 9 273 (1931) "Über Elementarakten mit zwei Quantensprünge ".

Experimental Requirements

- Non-linear effects → induced by high intensities
- Timescale of electronic dynamics \rightarrow attosecond pulses (1 as = 10^{-18} s)
- Coincidence spectroscopy → requires high repetition rates

Experiments will be held in **ELI ALPS, Hungary** [3] utilizing SYLOS laser system

- Intensities: ≤ 10¹⁷ W/cm²
- Pulse duration: < 500 as
- Repetition rate: 1 kHz



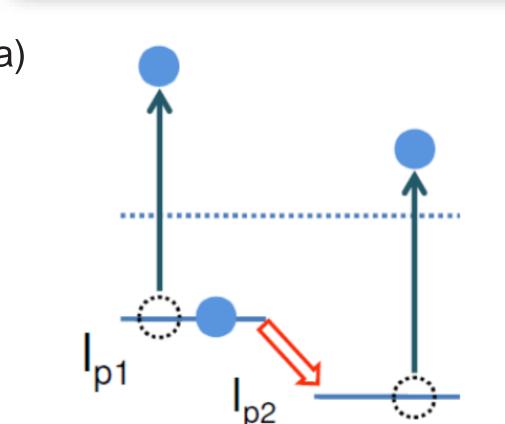
ELI- ALPS beamline

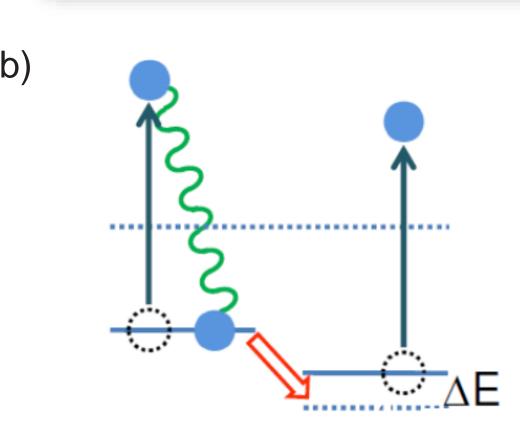
Investigating Electronic Dynamics

We choose a simple atomic system like He and we irradiate with **XUV** photons which acquire higher energy than the ionization potential. Let us have a closer look to the ionization pathways [1]:

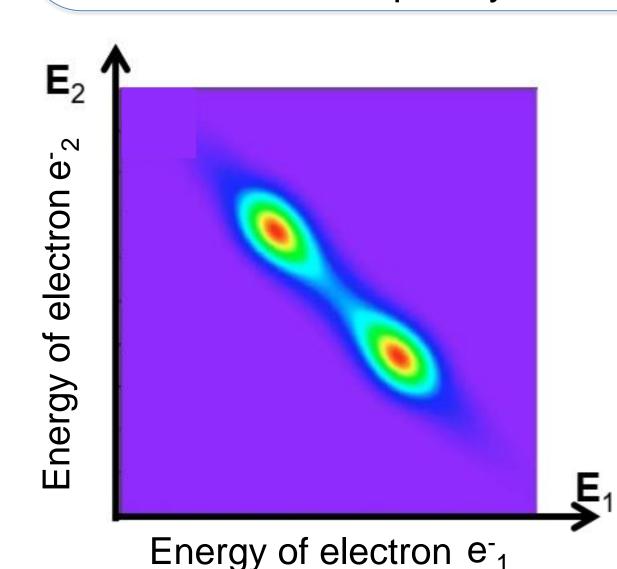
He +
$$\hbar\omega \to \text{He}^+(1\text{s}) + e_1^-$$
,
He⁺(1s) + $\hbar\omega \to \text{He}^{2+} + e_2^-$,

He +
$$\hbar\omega \rightarrow$$
 He⁺ $(nl) + e_1^-$,
He⁺ $(nl) + \hbar\omega \rightarrow$ He²⁺ $+ e_2^-$.





- a) A 1st photon ejects one electron (e⁻₁), leaving the ion in the ground state **He**⁺. At a later time, a 2nd photon is absorbed by **He**⁺ leading to a the emission of a second electron (e⁻₂).
- b) Here, electron correlation plays a role as the first electron leaving the atom can exchange energy promoting the second electron to an excited bound state, which is subsequently ionized by a 2nd photon.

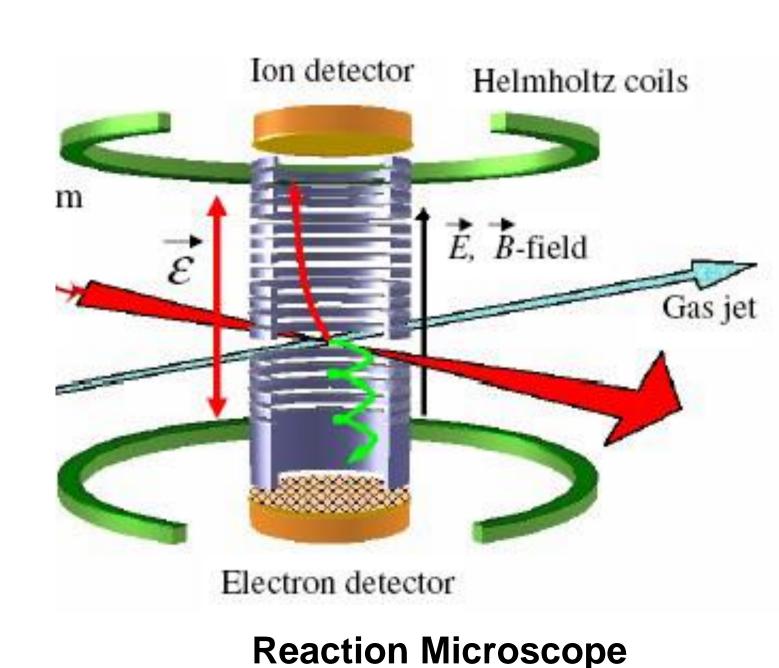


The plot depicts the acquired energy for the two emitted electrons [2]. When both electrons present similar energy electron correlation has also to be taken into account.

Considering the Heisenberg uncertainty, this correlation is oscillating on an attosecond timescale:

$$\Delta t = \frac{\hbar}{I_{p2} - I_{p1}} = 22 \ as$$

Reaction Microscope



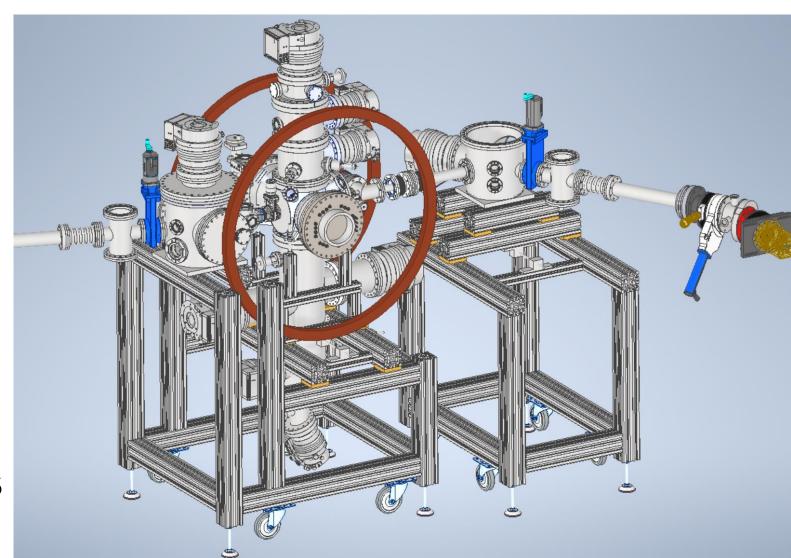
3D coincidence spectroscopy performed by means of the reaction microscope [4]

- Ionization of atoms/molecules by intense XUV pulses
 Accoleration of electrons/ions by
- Acceleration of electrons/ions by E-field
- Guidance of electrons by B-field
- Impact position → p_{x/v}
- Time of flight $\rightarrow p_7$

Upcoming Tasks

Mounting the setup

- Split-delay unit for pumpprobe experiments → characterization
- Setting up of an interlock system
- Acquisition software
- Measurements (testing the setup in Freiburg, Experimental measurements in ELI ALPS, Szeged)
- Data analysis



Sketch of the end station

Acknowledgments

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- 3. S. Kühn *et. al.,* J. Phys. B: At. Mol. Opt. Phys. **50** 132002, (2017).
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